### Ion Channels

STRUCTURE AND FUNCTION



#### Common properties of ION CHANNELS:

- GATING: mechanism that controls conformational transitions between open and closed state and therefore control OPENING and CLOSING of the channel
- **SELECTIVITY**: channel ability to select ion species that flows. Channels can be therefore classified by the selectivity properties.

The most direct way to study ion channels properties is by measuring ion fluxes or more precisely the electric current that flows in the channel. This is possible thanks to the patch-clamp technique introduced in the late 70's which allows to measure the current from single channel.





From the figure is clear that even in absence of stimulation, the channels can shift between two different levels:

#### CLOSE STATE; OPEN STATE

This is a common behavior of ion channels indicating that at least two conformational states exist: OPEN and CLOSE. The channel continuously shift between these two states



**OPEN STATE** 



The traces below also show a clear variability of open and close duration of the channel and therefore is not possible to predict how long a channel can stay in each of the functional state (open or close) or neither when the next transition will be

Stochastic events

The lows that describe these events are deducted from the probability distribution of several events number.



The kinetic state (the transitions) of a single channel, the stochastic activity of one event and the exponential distribution of the "duration histograms" can be explained by the **TRANSITION STATE THEORY** by Eyring which is premised on the somewhat tenuous assertion that "reactants (CHANNELS) rapidly thermalize with their surroundings until they reach the separatrix of the transition barrier, whereupon they inexorably turn to product."

In other words the channels undergo very rapid conformational changes which at the end determine shifts in the functional state (open or close states)

Transitions from OPEN to CLOSE state can be described:



A and  $\beta$  are speed constant and represent the number of transitions in the time unit. A and  $\beta$  can also be described in term of probability of transition in time frame t. This probability (that will not change in time if no conditions don't change) will be:  $\alpha$  dt  $\beta$  dt

Transitions from OPEN to CLOSE state can be described:



We can talk therefore about Probability of transitions episodes

In analogy with chemical reactions, the energetic profile of conformational changes is quite relevant.



Schematic representation of the energy profile free of conformational changes in ion channels. The diagram represents the free energy of different states involved in a transition: the initial state, activated complex, and final state. The equilibrium distribution between initial and final states depends on the relative value of their free energy (G0 and G1).

In analogy with chemical reactions, the energetic profile of conformations changes is quite relevant.



During the transitions between states, the channels pass through a transition state which is not favorable form the energetic point of view =  $G^*$ 

The bigger is  $\Delta G$ , the smaller is the probability of transition from Initial to Final state and thus the smaller will be  $\alpha$ 



This diagram represent the simplest case in which the channel present one initial and one final state.

02



In general channels have more complex energetic states.

**Open Probability**  $(p_o)$  is a very useful parameter commonly used to to measure channels activity and therefore the amount of ions crossing the membrane

 $P_{\rm o}$  describes the probability to find the channel open in a certain time fraction.

Probability to find the channel close

0 < p<sub>o</sub> < 1 Probability to find the channel open



 $tO_1 + tO_2 + tO_3$  $P_{o}$ = Tot time



Another way to calculate po is by amplitude histograms

$$P_o = \frac{AO}{AO + AC}$$

AC



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$$P_o = \frac{AO}{AO + AC}$$



The channels continuously shift from O to C states. If environmental condition (or experimental) change, the channels activity can change deeply. Po differs at different V



Po can be plotted in a Po/V curve





Po can be plotted in a Po/V curve  $Po = \frac{1}{1 + exp \left( \frac{z_g e (V-V_{1/2})}{kT} \right)}$ 

e = elementary charge k = Boltzman k T = Absolute T  $z_g$  = gating charge  $V_{1/2}$  = half activation V



The mechanism responsible of the voltage gating has been proposed initially by A.L. Hodgkin and A.F. Huxley. They observed that the variation of membrane permeability to Na+ and K+ were dependent on V changes. They hypothesized the **presence of V sensor in the form of charges** within the channel that are able to sense the voltage and move within the membrane thickness in response to V changes



As a consequence the Gating charges (zg) that move along the electric field within the membrane should be able to generate **transient currents**. These currents, although very small as compared with the ion currents through ion channels, **can be registered** strongly **supporting the hypothesis** of the voltage sensor



#### Molecular basis of Voltage sensor



Voltage gated- ion channels present a close structural homology. Each subunit of Kv or each of the 4 domains of the Na+ or Ca2+ V gated channels İS formed by 6 TM  $\alpha$  helix domains. S4 presents several basic aa: 4 to 7 repeated 3 positive residues followed by 2 hydrophobic aa residues

#### Molecular basis of Voltage sensor

These proposals presaged the idea that the S1-S4 segments serve as the voltage-sensing module while the S5 and S6 segments serve as the poreforming module and eventually led to the now-familiar sixtransmembrane-segment structural model for the domains of voltagegated Na+ channels



#### The Sliding Helix-Helical Screw Model for Voltage Sensor Function

How can an S4 segment containing four to seven positive charges (usually R) at three-residue intervals be stabilized in a transmembrane environment and move outward to translocate the gating charges across the membrane electric field?

Relying on thermodynamic and structural considerations, respectively, the ''sliding helix'' or ''helical screw'' models for voltage sensor function arrived at similar solutions to this conceptual problem.

	S2 TM	S4 ™	-
NaChBac K.AP	YRIDLVLLWI <u>F</u> TIEIAMRFLA YLVDLILVIILWADYAYRAYK	VLRILRVLRVLRAISVVP LFRLVRLLRFLRILLIIS	
K <sub>v</sub> 1.2	FIVETLCIIW <u>F</u> SFEFLVRFFA	ILRVIRLVRVFRIFKLSR	
Na <sub>v</sub> 1.2 I	KNVEYTFTGI <u>Y</u> TFESLIKILA	ALRTFRVLRALKTISVIP	
II	SVG <b>N</b> LVFTGI <u>F</u> TA <b>E</b> MFLKIIA	VLRSFRLLRVFKLAKSWP	
III	EYA <b>D</b> KVFTYI <u>F</u> IL <b>E</b> MLLKWVA	SLRTLRALRPLRALSRFE	
IV	YWINLVFIVL <u>F</u> TGECVLKLIS	VIRLARIGRILRLIKGAK	
	An 1 An2	R1 R2 R3 R4	

The charged residues in the S4 segments were proposed to form ion pairs with negatively charged amino acid residues in the neighboring \$1, S2, and/or S3 segments.

#### The Sliding Helix-Helical Screw Model for Voltage Sensor Function



In this configuration, the positively charged residues in the S4 segment are drawn inward by the electrostatic force of the negative internal resting membrane potential. Upon depolarization, this electrostatic force is relieved, and the S4 segments move outward along a spiral path such that each positively charged amino acid residue in the S4 segment makes a series of ion pairs with negative charges (Figure 1C).

#### The Sliding Helix-Helical Screw Model for Voltage Sensor Function

This proposed model for gating charge movement, hereinafter termed the **sliding-helix model** for brevity, makes four testable predictions:

- the positively charged residues in S4 serve as the gating charges
- the S4 segment is in a transmembrane position in both resting and activated states
- the S4 segment moves outward and rotates during activation
- the positive charges in the S4 segment form ion pairs sequentially with negative charges in neighboring transmembrane segments

Some of the Voltage-gated channels under a depolarizing stimulus, remain open just for a short period of time and then go back to a non permeable state eve in the presence of continuous depolarization = INACTIVATION. This is typical of Na<sup>+</sup> voltage-gated channels (Na<sup>+</sup><sub>v</sub>) and some K<sup>+</sup><sub>v</sub>.







If a second impulse is evoked immediately after, the current is much smaller but if the membrane is kept for enough time to negative potential, than depolarization will evoke again inward current with the same amplitude as the first one

Recovery from INACTIVATION requires membrane repolarization during which the channels pass from the inactive to Close state

From the kinetic point of view we can therefore describe the channel with the scheme:



Beside OPEN and CLOSED states these channels have a INACTIVE state in which the channels are just after the OPEN state.

The INACTIVE state is undistinguishable from the CLOSE state form the functional point of view since in both cases no Current permeate through the channel

On the other hand huge molecular differences exists between the INACTIVE and CLOSE state.

From the kinetic point of view we can therefore describe the channel with the scheme:



The passage to INACTIVE state at depolarizing potential is irreversible and the passage to CLOSE state requires hyperpolarizing condition of the plasma membrane.

Since the transition to INACTIVE state requires a depolarization step, we can state that it is a VOLTAGE DEPENDENT phenomenon such as the opening of the channel





To study this voltage dependence we can plot the peak amplitude of the current as a function of the V impulse imposed. These graph normally well described by a Boltzmann relation similarly to the one describing the Vdependence of the activation

$$Po = \frac{1}{1 + exp} = \frac{e (V - V_{1/2})}{kT}$$

From the molecular point of view, inactivation is due to intracellular component of the inactivating channels.

Both K+v Shaker and Na+v present the inactivating domain at the N terminal

= BALL AND CHAIN mechanism



#### Structure and function of Na+ Voltage-gated channel

